

DABCO-catalyzed regioselective cyclization reactions of β,γ -unsaturated α -ketophosphonates or β,γ -unsaturated α -ketoesters with allenic esters[†]

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Highly efficient DABCO-catalyzed [4 + 2] cycloaddition of β,γ -unsaturated α -ketophosphonates or β,γ -unsaturated α -ketoesters with allenic esters gives the corresponding highly functionalized tetrahydropyran and dihydropyran derivatives in good to excellent yields and moderate to good regioselectivities under mild conditions.

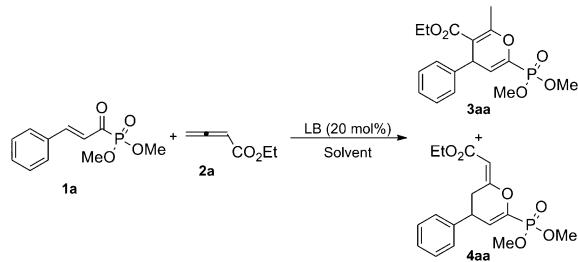
Introduction

Heterocycles are of great value in the design and discovery of new biologically active compounds.¹ The development of efficient processes to construct heterocycles, using metal-free catalysts, has been drawing much attention over the past decades.² Recently, nitrogen-containing Lewis base (LB) catalyzed cyclization reactions of allenotes have emerged as powerful synthetic tools in the rapid construction of cyclic molecular complexity.³ To our surprise, examples of β,γ -unsaturated α -ketophosphonates⁴ or β,γ -unsaturated α -ketoesters⁵ as the electrophiles were seldom mentioned in the construction of heterocycles. Herein, we wish to report a novel DABCO-catalyzed regioselective [4 + 2] cycloaddition of β,γ -unsaturated α -ketophosphonates or β,γ -unsaturated α -ketoesters with allenic esters to give the corresponding highly functionalized tetrahydropyran and dihydropyran derivatives, which are structural subunits in many natural products and biologically active molecules.⁶

Results and discussion

We initially utilized (*E*)-dimethyl cinnamoylphosphonate **1a** (0.1 mmol, 1.0 equiv) and ethyl 2,3-butadienoate **2a** (0.12 mmol, 1.2 equiv) as the substrates to investigate their cyclization behavior in tetrahydrofuran (THF) at room temperature in the presence of 20 mol% 1,4-diazabicyclo[2.2.2]octane (DABCO). It was found that the desired [4 + 2] cycloaddition reaction took place smoothly to give the corresponding cyclic products **3aa** and **4aa** in 87% combined yield but with low regioselectivity as the ratio of

Table 1 Optimization of the reaction conditions of (*E*)-dimethyl cinnamoylphosphonate **1a** and ethyl 2,3-butadienoate **2a**^a



Entry	LB	T/°C	Solvent	Yield (%) ^b	3aa : 4aa (%)^c
1	DABCO	rt	THF	87	2 : 1
2	DMAP	rt	THF	84	1 : 1
3	DBU	rt	THF	NR	—
4	Et ₃ N	rt	THF	NR	—
5	DIEA	rt	THF	NR	—
6	PPh ₃	rt	THF	NR	—
7	PBu ₃	rt	THF	NR	—
8	DABCO	rt	DCM	90	3 : 1
9	DABCO	rt	Et ₂ O	80	1 : 1
10	DABCO	rt	Dioxane	82	2 : 1
11	DABCO	rt	CH ₃ CN	81	2 : 1
12	DABCO	rt	Toluene	78	1 : 1
13	DABCO	rt	DCE	80	2 : 1
14	DABCO	rt	CHCl ₃	82	3 : 1
15	DABCO	rt	DMSO	trace	—
16	DABCO	rt	DMF	trace	—
17	DABCO	0	DCM	85	3 : 1
18	DABCO	-20	DCM	85	4 : 1
19	DABCO	-40	DCM	86	5 : 1
20	DABCO	-60	DCM	85	3 : 1

^a All reactions were carried out using **1a** (0.10 mmol) and **2a** (0.12 mmol) in solvent (1.00 mL) for 24 h. ^b Isolated combined yield. ^c Determined by ¹H NMR spectroscopic data, and these regioisomers can not be easily separated by column chromatography.

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[†] Electronic supplementary information (ESI) available: Experimental procedures, NMR charts for all compounds and X-ray crystal data of **6ea**. CCDC reference numbers 833190. For ESI and crystallographic data in CIF or other electronic format see DOI: 10.1039/c1ob06507f

3aa : 4aa was 2 : 1 within 24 h (Table 1, entry 1). Subsequently, we screened various nitrogen-containing Lewis base catalysts for this reaction, and the results are summarized in Table 1 (entries 2–7). 4-N,N-dimethylpyridine (DMAP) can also efficiently catalyze

this reaction to give the product mixture of **3aa** and **4aa** in 84% combined yield with a ratio of 1 : 1 (Table 1, entry 2). Other weak nucleophilic nitrogen-containing Lewis base catalysts, such as 1,8-diazabicyclo[5.4.0]-7-undecene (DBU), Et₃N and diisopropylethylamine (DIEA) did not catalyze this reaction (Table 1, entries 3–5). We next attempted to screen phosphane-containing Lewis base catalysts, such as PPh₃ and tributylphosphine (PBu₃), in this reaction, but it was found that no reactions occurred either (Table 1, entries 6 and 7). Of the catalysts examined, DABCO was found to be the best one. Using DABCO as the catalyst, various solvents were examined and dichloromethane (DCM) was found to be the solvent of choice, affording **3aa** and **4aa** in 90% combined yield with the ratio of 3 : 1 (Table 1, entries 8–16). Lowering the reaction temperature to 0 °C or –20 °C, similar results were obtained (Table 1, entries 17 and 18). At –40 °C, the corresponding cyclic adducts **3aa** and **4aa** were obtained in 86% combined yield along with a 5 : 1 ratio of **3aa** : **4aa** (Table 1, entry 19). Further reducing the reaction temperature did not improve the reaction outcomes (Table 1, entry 20). Thus, we have established the optimal reaction conditions for this reaction: using 20 mol % DABCO as a catalyst and DCM as a solvent to perform the reaction at –40 °C.

Under the optimized reaction conditions, the reaction generality was investigated by using various cinnamoylphosphonates **1** in the reaction with several allenic esters **2**, and the results of these experiments are summarized in Table 2. As can be seen from Table 2, changing the ester moiety of allenic esters **2** from OEt to OMe or O'Pr provided similar reaction outcomes, affording the desired products in good combined yields (up to 90%) along with moderate regioselectivities (**3** : **4** = 3 : 1) (Table 2, entries 2 and 3). However, we found that the R² substituent in the phosphonate moiety of

cinnamoylphosphonates **1** can significantly affect the reaction outcomes, as that R² group can improve the regioselectivity of the products (up to **3** : **4** = 10 : 1) if it had a sterically bulky group, such as O'Pr or O'Bu (Table 2, entries 4 and 5). As for substrates **1d**–**1h**, electron-withdrawing or electron-donating substituents at the *meta*- or *para*-positions of the benzene ring of **1** were equally well-tolerated in the reaction, giving the corresponding products **3** and **4** in good combined yields along with high regioselectivities (up to **3** : **4** > 20 : 1) (Table 2, entries 6–10). The substrate **1i**, in which R¹ is a 1-naphthyl group, was also able in this reaction to give the corresponding products in 85% combined yield along with good regioselectivity (**3ia** : **4ia** = 14 : 1) (Table 2, entry 11). When R¹ is an alkyl group (**1j**, R¹ = Me), the reaction also proceeded smoothly to give the desired products **3ja** and **4ja** in 75% combined yield but with moderate regioselectivity (**3ja** : **4ja** = 4 : 1) (Table 2, entry 12). The structures of **3** and **4** were determined by the 2-D NMR spectroscopic data (HMQC, HMBC, DEPT and NOESY spectra) of compounds **3ca** and **4ca** (see the Supporting Information for the details†).

Encouraged by the above results, β,γ-unsaturated α-ketoesters were also examined under the optimal reaction conditions. Initially, we utilized (E)-ethyl 2-oxo-4-phenylbut-3-enoate **5a** (0.10 mmol, 1.0 equiv) and ethyl 2,3-butadienoate **2a** (0.12 mmol, 1.2 equiv) as the substrates in DCM (2.00 mL, 0.05 M) at room temperature in the presence of 20 mol% DABCO. We were pleased to find that the reaction proceeded smoothly to give the desired products **6aa** and **7aa** in 82% combined yield along with good regioselectivity (**7aa** : **6aa** = 8 : 1) (Table 3, entry 1), which was contrary to the above results in Table 2. Reducing the concentration to 0.03 M by increasing the amount of solvent (DCM) employed to 3.00 mL can improve the reaction outcome, affording the corresponding products in 85% combined yield along with the 13 : 1 ratio of **7aa** : **6aa** (Table 3, entry 1). Inspired by this result, we next screened various nitrogen-containing Lewis base catalysts for this reaction in the optimized amount of solvent (3.00 mL, 0.03 M). DMAP can efficiently catalyze this reaction as well but with low regioselectivity. However, DBU and DIEA did not catalyze this reaction (Table 3, entries 2–4). Accordingly, DABCO was then used as the best catalyst for further investigation of solvent and temperature effects in this reaction. It was found that THF was the solvent of choice in comparison with those reactions carried out in other organic solvents, such as toluene, Et₂O, 1,2-dichloroethane (DCE), chloroform, 1,4-dioxane, CH₃CN, *N,N*-dimethylformamide (DMF) and dimethyl sulfoxide (DMSO) (Table 3, entries 5–13). The low temperature (–10 °C) can improve the combined yield of **6aa** and **7aa** (Table 3, entry 15), but further lowering the reaction temperature to –40 °C reduced the combined yield (Table 3, entry 16). Thus, we have identified the optimal reaction conditions for this reaction: using 20 mol% of DABCO as a catalyst and THF as a solvent to perform the reaction at –10 °C.

Having identified the optimal reaction conditions, we next set out to examine the scope and limitations of the [4 + 2] cycloaddition reaction catalyzed by DABCO using various β,γ-unsaturated α-ketoesters **5** with different substituents on the R⁴ group, and the results are summarized in Table 4. As can be seen from Table 4, when R⁴ is an aromatic group, whether they have electron-withdrawing or electron-donating substituents at the *ortho*-, *meta*- or *para*-positions on the benzene rings, the

Table 2 Substrate scope of the reactions of phosphonates **1** and allenic esters **2**^a

Entry	No	R ¹	R ²	R ³ (2)	Yield (%) ^b	No	3 : 4 ^c	
							3aa : 4aa	3 : 4
1	1a	C ₆ H ₅	OMe	Et (2a)	86	3aa : 4aa	5 : 1	
2	1a	C ₆ H ₅	OMe	'Pr (2b)	82	3ab : 4ab	3 : 1	
3	1a	C ₆ H ₅	OMe	Me (2c)	90	3ac : 4ac	3 : 1	
4	1b	C ₆ H ₅	O'Bu	Et (2a)	70	3ba : 4ba	9 : 1	
5	1c	C ₆ H ₅	O'Pr	Et (2a)	85	3ca : 4ca	10 : 1	
6	1d	4-ClC ₆ H ₄	O'Pr	Et (2a)	87	3da : 4da	10 : 1	
7	1e	4-NO ₂ C ₆ H ₄	O'Pr	Et (2a)	80	3ea : 4ea	>20 : 1	
8	1f	4-BrC ₆ H ₄	O'Pr	Et (2a)	85	3fa : 4fa	13 : 1	
9	1g	3-MeC ₆ H ₄	O'Pr	Et (2a)	85	3ga : 4ga	10 : 1	
10	1h	4-MeOC ₆ H ₄	O'Pr	Et (2a)	75	3ha : 4ha	8 : 1	
11	1i	1-naphthyl	O'Pr	Et (2a)	85	3ia : 4ia	14 : 1	
12	1j	Me	O'Pr	Et (2a)	75	3ja : 4ja	4 : 1	

^a All reactions were carried out using **1** (0.10 mmol) and **2** (0.12 mmol) in DCM (1.00 mL) at –40 °C for 24 h. ^b Isolated combined yield. ^c Determined by ¹H NMR spectroscopic data, and these regioisomers cannot be easily separated by column chromatography.

Table 3 Optimization of the reaction conditions of (*E*)-ethyl 2-oxo-4-phenylbut-3-enoate **5a** and ethyl 2,3-butadienoate **2a**^a

Entry	LB	T/°C	Solvent	Yield (%) ^b	7aa : 6aa (%)^c
1	DABCO	rt	DCM	85 (82) ^d	13:1 (8:1) ^d
2	DMAP	rt	DCM	80	2:1
3	DIEA	rt	DCM	NR	—
4	DBU	rt	DCM	NR	—
5	DABCO	rt	THF	90	>20:1
6	DABCO	rt	Toluene	75	5:1
7	DABCO	rt	Et ₂ O	88	13:1
8	DABCO	rt	DCE	86	8:1
9	DABCO	rt	CHCl ₃	90	6:1
10	DABCO	rt	Dioxane	89	14:1
11	DABCO	rt	CH ₃ CN	50	>20:1
12	DABCO	rt	DMSO	trace	—
13	DABCO	rt	DMF	trace	—
14	DABCO	0	THF	91	>20:1
15	DABCO	-10	THF	95	>20:1
16	DABCO	-40	THF	92	>20:1

^a All reactions were carried out using **5a** (0.10 mmol) and **2a** (0.12 mmol) in solvent (3.00 mL, 0.03 M) for 24 h. ^b Isolated combined yield. ^c Determined by ¹H NMR spectroscopic data, and these regioisomers can not be easily separated by column chromatography. ^d The reaction was carried out in DCM (2.00 mL, 0.05 M).

Table 4 Substrate scope of the reactions of β,γ -unsaturated α -ketoesters **5** and ethyl 2,3-butadienoate **2a**^a

Entry	No	R ¹	Yield (%) ^b	No	7 : 6 (%)^c
1	5a	C ₆ H ₅	95	7aa : 6aa	>20:1
2	5b	4-ClC ₆ H ₄	86	7ba : 6ba	14:1
3	5c	4-BrC ₆ H ₄	87	7ca : 6ca	10:1
4	5d	3-BrC ₆ H ₄	83	7da : 6da	8:1
5	5e	2-BrC ₆ H ₄	85	7ea : 6ea	8:1
6	5f	2,4-Cl ₂ C ₆ H ₃	87	7fa : 6fa	7:1
7	5g	4-MeC ₆ H ₄	85	7ga : 6ga	12:1
8	5h	2-furyl	60	7ha : 6ha	3:1
9	5i	2-thiophene	57	7ia : 6ia	6:1
10	5j	2-naphthyl	82	7ja : 6ja	8:1
11	5k	cyclopropyl	90	7ka : 6ka	>20:1

^a All reactions were carried out using **5** (0.10 mmol) and **2a** (0.12 mmol) in solvent (3.00 mL, 0.03 M) for 24 h. ^b Isolated combined yield. ^c Determined by ¹H NMR spectroscopic data, and these regioisomers can not be easily separated by column chromatography.

reactions proceeded smoothly to give the desired products in good yields along with good regioselectivities (Table 4, entries 2–7). Heterocyclic substrates **5h** and **5i** were also suitable in this reaction to give the corresponding products in moderate combined yields and regioselectivities (Table 4, entries 8 and 9). The substrate **5j**, in which R⁴ is a 2-naphthyl group, was also tolerable in this reaction to give the corresponding products in 82% combined yield along with good regioselectivity (**7ja : 6ja** = 8:1) (Table 4, entry 10). For substrate **5k**, in which R⁴ is a cyclopropyl group, the reaction also proceeded smoothly to give the desired products in high combined yield (up to 90%) along with high regioselectivity (up to **7ka : 6ka** = >20:1). The structure of **6ea** was unambiguously determined by X-ray diffraction. The ORTEP drawing is shown in Fig. 1 and its CIF data are summarized in the Supporting Information.^{†7} The structure of **7** and the *E*-configuration of the double bond were assigned by the 2-D NMR spectroscopic data (HMQC, HMBC, DEPT and NOESY spectra) of compound **7aa** (see the Supporting Information for the details^{†7}).

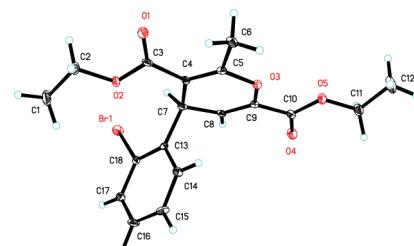
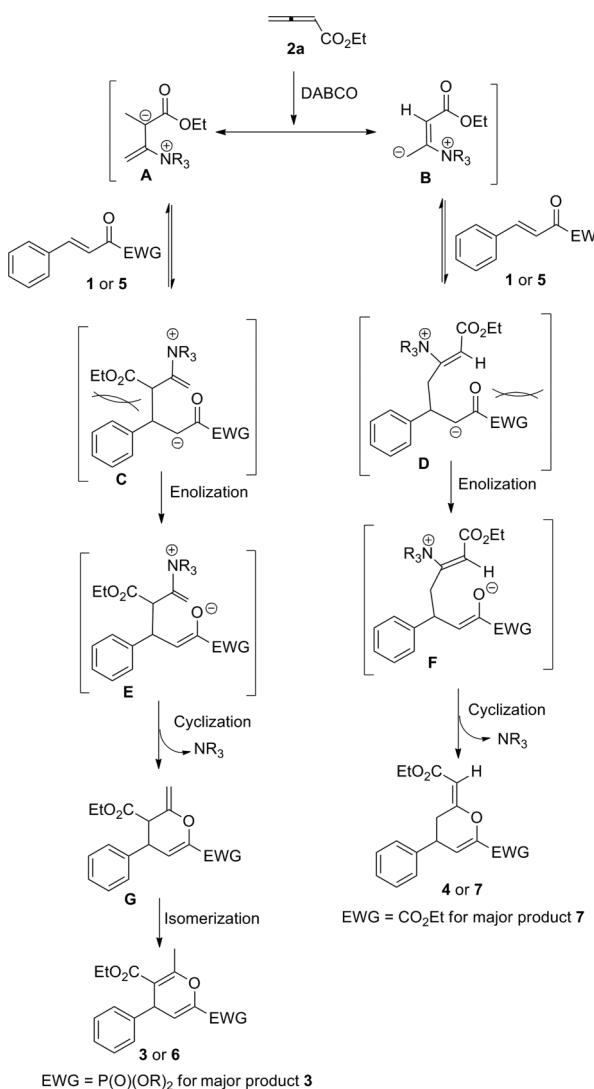


Fig. 1 An ORTEP drawing of **6ea**.

The mechanism for the reactions has not been unequivocally established, but one reasonable explanation is shown in Scheme 1 based on earlier reports and our own investigations. Addition of DABCO to allenate **2a** delivers zwitterionic intermediate **A**, which coexists with its resonance form **B**.^{3a,3c} The zwitterionic intermediate **A** reacts with **1** or **5** to give intermediate **C**, which undergoes enolization to give intermediate **E**. Subsequent cyclization produces cyclic product **G** and regenerates the DABCO catalyst. Finally, **G** isomerizes to give the more stable adducts **3** or **6**. At the same time, the zwitterionic intermediate **B** can also react with **1** or **5** to give intermediate **D**, which undergoes enolization to give intermediate **F**. Subsequent cyclization produces cyclic product **4** or **7** and regenerates the DABCO catalyst. We also assumed that the major products in the reaction might be mainly due to the steric interaction between intermediates **C** and **D**. When EWG = PO(O*i*Pr)₂, which is sterically more bulky than an aromatic ring, the steric repulsion between CO₂Et and EWG is larger than that of CO₂Et and an aromatic ring. Therefore, intermediate **C** is more stable than intermediate **D**, affording **3** as the major product. When EWG = CO₂Et, which is sterically smaller than aromatic ring, the steric repulsion between CO₂Et and EWG is less than that of CO₂Et and aromatic ring. Therefore, intermediate **D** is more stable than intermediate **C**, leading to **7** as the major product.

In conclusion, we have established a novel DABCO-catalyzed [4 + 2] cycloaddition of β,γ -unsaturated α -ketophosphonates or β,γ -unsaturated α -ketoesters with allenic esters to give the corresponding highly functionalized tetrahydropyran and dihydropyran derivatives in good to excellent yields and moderate



Scheme 1 A plausible reaction mechanism.

to good regioselectivities under mild conditions. The obtained multiple functionalized tetrahydropyran and dihydropyran derivatives are useful building blocks in the organic synthesis of biologically useful compounds. A plausible reaction mechanism has been also proposed on the basis of previous literature and our own investigations. Efforts are in progress to elucidate further mechanistic details of these reactions and to understand their scope and limitations.

Experimental section

General remarks

^1H NMR spectra were recorded on a Bruker AM-300 or AM-400 spectrometer for solution in CDCl_3 with tetramethylsilane (TMS) as internal standard; J -values are in Hz. Mass spectra were recorded with a HP-5989 instrument. All of the compounds reported in this paper gave satisfactory HRMS analytic data. Melting points were determined on a digital melting point apparatus and temperatures were uncorrected. Infrared spectra were recorded on a Perkin-Elmer PE-983 spectrometer with absorption

in cm^{-1} . THF, toluene and Et_2O were distilled from sodium (Na) under argon (Ar) atmosphere. CH_3CN , 1,2-dichloroethane and dichloromethane were distilled from CaH_2 under argon (Ar) atmosphere. Commercially obtained reagents were used without further purification. All reactions were monitored by TLC with Huanghai GF254 silica gel coated plates. Flash column chromatography was carried out using 300–400 mesh silica gel at increased pressure. All the β,γ -unsaturated α -ketophosphonates and β,γ -unsaturated α -ketoesters were prepared according to the literature.^{4b,8}

General procedure for the preparation of **3** and **4** from the reaction of **1a** with **2a** using **3aa** and **4aa** as an example in the presence of DABCO

To a mixture of **1a** (0.10 mmol, 24.0 mg), **2a** (0.12 mmol, 13.6 μL) and DABCO (2.2 mg, 0.02 mmol) was added 2.0 mL of dichloromethane at -40°C . The reaction solution was monitored by TLC. After the reaction was complete, the solution was concentrated under reduced pressure and the residue was further purified by silica gel column chromatography ($\text{EtOAc}/\text{PE} = 1/6$) to give the target products **3aa** and **4aa**.

Ethyl 6-(dimethoxyphosphoryl)-2-methyl-4-phenyl-4*H*-pyran-3-carboxylate **3aa.** A colorless oil (16.7 mg, 70%); ^1H NMR (400 MHz, CDCl_3 , TMS) δ 1.09 (t, $J = 7.2$ Hz, 3H), 2.39 (s, 3H), 3.78 (d, $J = 11.2$ Hz, 3H), 3.80 (d, $J = 11.2$ Hz, 3H), 3.96–4.07 (m, 2H), 4.45 (d, $J = 5.2$ Hz, 1H), 6.12 (dd, $J = 10.0$ Hz, 5.2 Hz, 1H), 7.19–7.22 (m, 3H), 7.29–7.31 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 13.8, 19.0, 38.1 (d, $J = 10.4$ Hz), 53.3 (d, $J = 6.7$ Hz), 53.4 (d, $J = 5.9$ Hz), 60.1, 104.8, 122.0 (d, $J = 20.9$ Hz), 126.9, 127.8, 128.5, 137.4, 139.8, 144.0 (d, $J = 2.2$ Hz), 159.7 (d, $J = 8.2$ Hz), 166.6; ^{31}P NMR (161.93 MHz, CDCl_3 , 85% H_3PO_4) δ 9.734; IR (CH_3Cl_2) ν 2966, 2902, 1715, 1659, 1473, 1373, 1260, 1176, 1105, 1026, 947, 800, 741, 700 cm^{-1} ; MS (ESI) m/z 353.0 ($\text{M}+\text{H}^+$). HRMS (ESI) Calcd for $\text{C}_{17}\text{H}_{22}\text{O}_6\text{P}$ requires ($\text{M}+\text{H}^+$): 353.1149, found: 353.1156.

(E)-Ethyl 2-(6-(dimethoxyphosphoryl)-4-phenyl-3,4-dihydro-2*H*-pyran-2-ylidene)acetate **4aa.** A colorless oil (5.5 mg, 16%); ^1H NMR (400 MHz, CDCl_3 , TMS) δ 1.23 (t, $J = 7.2$ Hz, 3H), 3.10 (dd, $J = 10.4$ Hz, 6.8 Hz, 1H), 3.67–3.73 (m, 2H), 3.84 (d, $J = 11.2$ Hz, 3H), 3.85 (d, $J = 11.2$ Hz, 3H), 4.08–4.14 (m, 2H), 5.67 (s, 1H), 6.28 (dd, $J = 10.4$ Hz, 2.8 Hz, 1H), 7.19–7.22 (m, 2H), 7.26–7.28 (m, 1H), 7.31–7.35 (m, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 14.1, 29.8, 35.6 (d, $J = 11.9$ Hz), 53.39 (d, $J = 3.7$ Hz), 53.40 (d, $J = 6.2$ Hz), 59.8, 100.8, 122.0 (d, $J = 19.3$ Hz), 127.0, 127.2, 128.8, 141.0 (d, $J = 11.2$ Hz), 143.4, 164.6 (d, $J = 9.7$ Hz), 166.6; ^{31}P NMR (161.93 MHz, CDCl_3 , 85% H_3PO_4) δ 9.180; IR (neat) ν 2966, 2902, 1712, 1656, 1494, 1449, 1374, 1261, 1172, 1111, 1026, 824, 764, 700 cm^{-1} ; MS (ESI) m/z 353.1 ($\text{M}+\text{H}^+$). HRMS (ESI) Calcd for $\text{C}_{17}\text{H}_{22}\text{O}_6\text{P}$ requires ($\text{M}+\text{H}^+$): 353.1149, found: 353.1159.

Isopropyl 6-(dimethoxyphosphoryl)-2-methyl-4-phenyl-4*H*-pyran-3-carboxylate **3ab.** A slightly yellow liquid (22.0 mg, 60%); ^1H NMR (CDCl_3 , 400 MHz, TMS) δ 0.91 (d, $J = 6.0$ Hz, 3H), 1.16 (d, $J = 6.0$ Hz, 3H), 2.39 (s, 3H), 3.78 (d, $J = 11.2$ Hz, 3H), 3.81 (d, $J = 11.2$ Hz, 3H), 4.44 (d, $J = 4.8$ Hz, 1H), 4.88 (sept, $J = 6.0$ Hz, 1H), 6.11 (dd, $J = 10.4$ Hz, 4.8 Hz, 1H), 7.18–7.31 (m, 5H); ^{13}C NMR (100 MHz, CDCl_3) δ 19.0, 21.3, 21.8, 38.3 (d, $J = 10.6$ Hz), 53.29 (d, $J = 6.0$ Hz), 53.35 (d, $J = 6.0$ Hz), 67.6, 105.0,

(ESI) m/z 439.3 ($M+H^+$); HRMS (ESI) Calcd for $C_{22}H_{31}O_7PNa$ requires ($M+Na^+$): 461.1700, found: 461.1694.

Ethyl 6-(diisopropoxyphosphoryl)-2-methyl-4-(naphthalen-1-yl)-4H-pyran-3-carboxylate 3ia. A slightly yellow liquid (36.2 mg, 79%); 1H NMR ($CDCl_3$, 400 MHz, TMS) δ 0.74 (t, J = 7.2 Hz, 3H), 1.25 (d, J = 6.4 Hz, 3H), 1.29 (d, J = 5.2 Hz, 3H), 1.30 (d, J = 6.0 Hz, 3H), 1.36 (d, J = 6.4 Hz, 3H), 2.49 (s, 3H), 3.85 (q, J = 7.2 Hz, 2H), 4.53–4.62 (m, 1H), 4.65–4.73 (m, 1H), 5.34 (d, J = 4.8 Hz, 1H), 6.21 (dd, J = 10.4 Hz, 4.8 Hz, 1H), 7.32 (d, J = 6.8 Hz, 1H), 7.42 (t, J = 8.0 Hz, 1H), 7.47–7.56 (m, 2H), 7.71 (d, J = 8.0 Hz, 1H), 7.86 (d, J = 8.0 Hz, 1H), 8.19 (d, J = 8.0 Hz, 1H); ^{13}C NMR ($CDCl_3$, 100 MHz, TMS) δ 13.6, 18.9, 23.5 (d, J = 4.4 Hz), 23.6 (d, J = 5.2 Hz), 23.87 (d, J = 4.4 Hz), 23.91 (d, J = 4.4 Hz), 33.1 (d, J = 10.5 Hz), 60.0, 71.66 (d, J = 5.2 Hz), 71.72 (d, J = 6.0 Hz), 104.3 (d, J = 1.5 Hz), 120.2 (d, J = 21.1 Hz), 122.5, 125.60, 125.61, 126.3, 127.3, 128.7, 130.4, 133.7, 139.5, 140.8 (d, J = 2.2 Hz), 141.8, 160.4 (d, J = 8.2 Hz), 166.8; ^{31}P NMR ($CDCl_3$, 161.93 MHz, 85% H_3PO_4) δ 4.637; IR (CH_2Cl_2) ν 2978, 2906, 1713, 1657, 1375, 1351, 1261, 1114, 1048, 1014, 803, 739, 703 cm $^{-1}$; MS (ESI) m/z 347.2 ($M+H^+$); HRMS (ESI) Calcd for $C_{25}H_{31}O_6PNa$ requires ($M+Na^+$): 369.1438, found: 369.1421.

General procedure for the preparation of 6 and 7 from the reaction of 5a with 2a using 6aa and 7aa as an example in the presence of DABCO

To a mixture of **1a** (0.10 mmol, 20.4 mg), **2a** (0.12 mmol, 13.6 μ L) and DABCO (2.2 mg, 0.02 mmol) was added 3.0 mL of THF at –10 °C. The reaction solution was monitored by TLC. After the reaction was complete, the solution was concentrated under reduced pressure and the residue was further purified by silica gel column chromatography (EtOAc/PE = 1/6) to give the target products **6aa** and **7aa**.

Diethyl 6-methyl-4-phenyl-4H-pyran-2,5-dicarboxylate 6aa. A slightly yellow liquid (2.7 mg, 7%); 1H NMR ($CDCl_3$, 400 MHz, TMS) δ 1.08 (t, J = 7.2 Hz, 3H), 1.30 (t, J = 7.2 Hz, 3H), 2.44 (s, 3H), 3.99–4.06 (m, 2H), 4.21–4.30 (m, 2H), 4.51 (d, J = 5.2 Hz, 1H), 6.26 (d, J = 5.2 Hz, 1H), 7.20–7.24 (m, 3H), 7.29–7.33 (m, 2H); ^{13}C NMR ($CDCl_3$, 100 MHz, TMS) δ 13.9, 14.1, 19.1, 38.9, 60.2, 61.5, 104.6, 116.2, 127.0, 128.1, 128.6, 138.9, 144.4, 159.8, 161.1, 166.9; IR (CH_2Cl_2) ν 2980, 2903, 1736, 1714, 1658, 1448, 1374, 1260, 1118, 1018, 860, 802, 761, 739, 700 cm $^{-1}$; MS (ESI) m/z 317.2 ($M+H^+$); HRMS (ESI) Calcd for $C_{18}H_{20}O_5Na$ requires ($M+Na^+$): 339.1203, found: 339.1206.

(E)-Ethyl 2-(2-ethoxy-2-oxoethylidene)-4-phenyl-3,4-dihydro-2H-pyran-6-carboxylate 7aa. A slightly yellow liquid (30.0 mg, 95%); 1H NMR ($CDCl_3$, 400 MHz, TMS) δ 1.23 (t, J = 7.2 Hz, 3H), 1.34 (t, J = 7.2 Hz, 3H), 3.11 (dd, J = 8.0 Hz, 4.8 Hz, 1H), 3.68–3.72 (m, 2H), 4.07–4.14 (m, 2H), 4.28–4.34 (m, 2H), 5.75 (s, 1H), 6.45 (d, J = 4.8 Hz, 1H), 7.20–7.22 (m, 2H), 7.26–7.28 (m, 1H), 7.31–7.35 (m, 2H); ^{13}C NMR ($CDCl_3$, 100 MHz, TMS) δ 14.1, 14.2, 29.7, 35.8, 59.8, 61.6, 101.0, 116.5, 127.2, 128.8, 141.3, 141.7, 161.1, 164.6, 166.9; IR (CH_2Cl_2) ν 2981, 2903, 2875, 1736, 1711, 1660, 1493, 1373, 1258, 1170, 1118, 1046, 1015, 821, 760 cm $^{-1}$; MS (ESI) m/z 317.2 ($M+H^+$). HRMS (ESI) Calcd for $C_{18}H_{20}O_5$ requires ($M+Na^+$): 339.1203, found: 339.1206.

Diethyl 4-(4-chlorophenyl)-6-methyl-4H-pyran-2,5-dicarboxylate 6ba. A slightly yellow liquid (2.0 mg, 6%); 1H NMR ($CDCl_3$, 400 MHz, TMS) δ 1.11 (t, J = 7.2 Hz, 3H), 1.31 (t, J = 7.2 Hz, 3H), 2.44 (s, 3H), 4.00–4.08 (m, 2H), 4.22–4.31 (m, 2H), 4.49 (d, J = 4.8 Hz, 1H), 6.21 (d, J = 4.8 Hz, 1H), 7.16 (d, J = 8.8 Hz, 2H), 7.27 (d, J = 8.8 Hz, 2H); ^{13}C NMR ($CDCl_3$, 100 MHz, TMS) δ 14.00, 14.04, 19.2, 38.4, 60.3, 61.6, 104.3, 115.5, 128.7, 129.4, 132.8, 139.1, 142.9, 160.1, 161.0, 166.7; IR (CH_2Cl_2) ν 2981, 2904, 1739, 1715, 1629, 1488, 1372, 1327, 1268, 1106, 1049, 1015, 946, 805, 739 cm $^{-1}$; MS (ESI) m/z 373.2 ($M+Na^+$); HRMS (ESI) Calcd for $C_{18}H_{19}ClNaO_5$ requires ($M+Na^+$): 373.0816, found: 373.0813.

(E)-Ethyl 4-(4-chlorophenyl)-2-(2-ethoxy-2-oxoethylidene)-3,4-dihydro-2H-pyran-6-carboxylate 7ba. A slightly yellow liquid (28.1 mg, 80%); 1H NMR ($CDCl_3$, 400 MHz, TMS) δ 1.24 (t, J = 7.2 Hz, 3H), 1.35 (t, J = 7.2 Hz, 3H), 3.18 (dd, J = 15.2 Hz, 7.6 Hz, 1H), 3.57 (dd, J = 15.2 Hz, 6.0 Hz, 1H), 3.68–3.72 (m, 1H), 4.08–4.14 (m, 2H), 4.29–4.35 (m, 2H), 5.75 (s, 1H), 6.40 (d,

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